## Tenneco Minerals A Tenneco Company

M/053/005

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MAY 1 8 1988

TENNECO

DIV. OIL, GAS, MINING

P.O. Box 2650

St. George, Utah 84770

May 16, 1988

Holland Shepherd Utah Dept. of Natural Resources Div. of Oil, Gas & Mining 3 Triad Center, Suite 350 Salt Lake City, Utah 84180-1203

Dear Mr. Shepherd;

I believe we now have enough information to demonstrate that acid mine drainage will not be a problem at the Goldstrike Mine. At Dave Whams request we estimated the tonnage and acid potential for sulfide ores and waste in the projected pits. The results of that study indicated 336,790 tons of sulfide material with ore grade gold values. Another 155,190 tons of sulfide had lower than ore grade gold values (in other words waste). The total amount of sulfide material identified at the mine is then 491,980 tons. We analyzed five samples of sulfide material using the EPA-600/2-78-054 procedure recomended by Dave Wham. (See attached reports from ACZ Inc.) As you can see the average acid potential was a -39.

Since the acid forming potential exceeded the -5 threshold mentioned by Dave Wham, we analyzed the pits for potential neutralizing agents. The Hamburg Pit which contains all the sulfide material also has 2,073,000 tons of limestone. Four samples of this material were analyzed using the same EPA procedure. The analyses results averaged +603. Thus each ton of limestone could neutralize more than 15 times its own weight in acid or 31,095,000 tons of the identified acid material.

As was pointed out earlier the sulfide waste amounted to 155,190 tons. Operationally, it would not be difficult to designate a specific location on the waste dumps for disposal of both acid forming and acid neutralizing materials. The materials will be mixed by the dumping action of the trucks. Further much of the leach pad area will be built on these waste dumps. One specification for Leach pad foundations is installation of 10 feet of engineered fill on top of end dump fill. Tenneco Mineralshas no problem with committing to not including sulfide materials in any part of an engineered fill. Therefore the sulfide waste will also be buried and oxygen will not readily come into contact with the sulfides.

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The ore portion of the sulfide material will be placed on leach pads just like regular ore. At this juncture all that is certain is that the sulfide ore grade material will be placed on a separate pad with a liner comparable to that used for regular leaching. The solutions generated from this separate leach pad will also be kept separate from normal leach liquors.

I hope this dicussion allays yours concerns regarding the sulfide materials in the Goldstrike pits. I await your reply.

Sincerely Yours,

K.W. Grubaugh P.E.

Pit Operations Supervisor

KWG/kg

cc:M.D. Stairwalt
M. Litus

*	******	*******	******	*******	******
*	Goldstrike M	ine	Da	te: 5/15/88	*
*	Acid/Base Spo	oil Analysis			*
*					*
*	******	*******	******	******	*****
*	Sample Name	Sulfur Ne	utralization	Acid-Base	*
*	particular states and an experience of the second states and a second state of the second states and the second states are second states as the second states are sec	Total	Potential	Potential	*
*		%	% CaCO3	Tons CaCO3/	1000 tons *
*	******	******	**********		
*	Acid Forming	Materials			*
*	88-51/0142	1.47%	<1	-46	*
*	88-51/0143	1.25%	1	-38	*
*	88-51/0144	0.82%	2 .	-24	*
*	88-51/0145	1.20%	3	-35	*
	88-51/0146		<1	-52	*
*					*
*	Total	6.40%	2	-195	*
*	Average	1.28%	0.4	-39	*
*					*
*	Acid Neutral	izing Materi	als		*
*	88-\$1/0352		52.80%	525	*
*	88-\$1/0353	0.04%	66.80%	667	*
*	88-\$1/0354	0.03%	67.20%	671	*
*	88-\$1/0355	0.03%	55.00%	549	*
*					*
*	Total	0.21%	241.80%	2412	*
*	Average	0.05%	60.45%	603	*
*					*
*	Available Ne	utralization	At Goldstrike		*
*	603/39=	15.46			*
*	In other wor	ds 1000 tons	s limestone will	neutralize 15	460 tons *
*	of acid form	ing material	S.		*
*					*
*					*
*					*
*					*
*	******	*****	*******	********	******

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## ACZ INC./LABORATORY DIVISION SOILS ANALYSES REPORT

Client:

Tenneco Minerals

P.O. Box 2650

St. George, Utah 84770 Ms. Julie Lewis

Attn:

Report Date: Date Received:

05/16/88 05/05/88

LAB NO.	SAMPLE I.D.	Sulfur Total %	Neutralization Potential % CaCO3	Acid-Base Potential Tons CaCO3/1000 tons	
88-SI/0352	HB-1 (PC)	0.11	52.8	525	
88-SI/0353	HB-2 (TL1)	0.04	66.8	667	
88-SI/0354	HB-3 (MR)	0.03	67.2	671	
88-SI/0355	MZ-1 (PC)	0.03	55.0	549	

Ralph V. Poulsen

Ralph V. Poulsen, Laboratory Director

	CLIENT Tennoco		#24.00 / sample			Tenneco Minerals ROB 2650 St. George Ut 84770 Ms. Julie Lewis	SHELL			1	
	SAMPLE	PLE Sulfue, Total		Neutralization Acid-Base Potential Potential		Poleutial	SAMPLE IO				
-	TARGET STATE OF THE STATE OF TH			1 % Cal		Tons Cacos/1000 ton				03/22/1988	
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	/43	1	25		!	-38	86-84-115'-160'				16:42
-	144	144 82 2				- 24	86-85-80. 90. 92, 102.				
1	145	1	1 20 3 -35 86-88-95'-130' 1 66 < 1 '-52 86-132 75'-105'								-0
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